

# X-ray Photoelectron Spectroscopy (XPS) or Electron Spectroscopy for Chemical Analysis (ESCA)



## SPECIFICATIONS

- ▶ Qualitative **elemental analysis**: all elements can be detected except H and He
- ▶ Average detection level : **0.1 - 0.5 atomic%**
- ▶ **Quantitative** analysis: precision 2-5%, accuracy 20 %
- ▶ **Surface analysis** : analysis depth < 100 Å (about 30 to 50 Å)
- ▶ **Chemical analysis** : type of bonds, oxide/metal rate...
- ▶ Spot size (for our device): 10 μm to 200 μm
- ▶ **Non destructive...** except in case of **concentration profiles**
- ▶ **Conductive and insulating samples**
- ▶ **Quantitative depth profiling**
- ▶ **Quantitative chemical imaging**
- ▶ Analysis under **ultrahigh vacuum** ( $10^{-9}$  -  $10^{-10}$  torr)

## PRINCIPLE

The surface of the sample is bombarded by a monochromatic X-ray beam. X-ray photons are absorbed by the material, the energy transferred,  $E_x$ , is partly used to excite core levels electrons:  $E_x = E_i + E_c$ , ( $E_i$  binding energy and  $E_c$  kinetic energy of the excited electron).

The detection consists of a kinetic energy filtering of emitted electrons. The **XPS spectrum** is described by a **succession of peaks which correspond to a given EI** (for example, C 1s peak = excitation of carbon 1s level electrons), that's why it is possible to perform **elemental analyses**.

There are few interferences between peaks of the different elements. Even when a peak can correspond to several elements, there is, in general, a way to satisfy doubts by studying the whole elemental spectrum: presence/absence of peaks, relative intensity of characteristic peaks of the suspected elements, study of the Auger peaks, ... The signal under each peak of element A is proportional to the number of A atoms, so this **analysis** can be **quantitative**. Finally, the signal intensity as a function of sample thickness  $t$  is damped by  $\exp(-t/\lambda)$  with  $\lambda$ , the mean free path of electrons in material: the deepest you go in the sample, the less the ejected electrons have a probability to be detected, i.e., the weakest is their contribution to total signal.

Thus 70 % of detected signal comes from the first  $\lambda$ , and beyond  $3\lambda$ , the contribution is negligible.  $\lambda$  being about 10 to 20 Å, the thickness analyzed is about 30 to 50 Å, that's why XPS is a **surface analysis** technique.

